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C-Glucosylflavonoid biosynthesis from 2-hydroxynaringenin by Desmodium uncinatum (Jacq.) (Fabaceae)

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ABSTRACT

[2',3',5',6'- 2 H₄]-2-Hydroxynaringenin is synthesised and incubated with commercially available UDP-glucose and the crude protein extract from Desmoduim uncinatum leaves. The organic extract produces isotopically labelled [2',3',5',6'- 2H_4]-vitexin and [2',3',5',6'- 2H_4]-isovitexin. Repeating the experiment with denatured protein or replacing the 2-hydroxynaringenin with $[2',3',5',6'-^2H_4]$ -apigenin or $[2',3',5',6'-^2H_4]$ naringenin results in no observable incorporation. 2-Hydroxynaringenin is therefore the substrate for Cglucosylflavonoid biosynthesis in D. uncinatum.

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Desmodium uncinatum (Jacq.) is used as an intercrop in subsistence farming of maize (Zea mays) in East Africa where it has demonstrated effectiveness in the suppression of parasitism by the parasitic plant Striga hermonthica (Del.) Benth., a weed that can devastate crops in the region. $1-4$ The effect was shown to be allelopathic⁵ and during investigations into the mechanism, we have discovered C-glycosylflavonoids, in particular the di-C-glycosylflavonoid, 6-C-a-L-arabinopyranosyl-8-C-b-D-glucopyranosylapigenin (isoschaftoside), in the root extract and root exudates of D. uncinatum that affect the early stages of Striga development.⁶

Elucidating the biosynthetic pathway for this class of compound and characterising the enzymes that control C-glycosylflavonoid biosynthesis provides the potential for transferring the mechanism for Striga protection from this cattle forage legume into an edible crop legume, 6 and C-glycosyltransferase (CGT) activity is a key role. Previous biosynthetic studies have shown incorporation of radiolabelled naringenin and p-coumaric acid but not apigenin into Cglucosylflavones by Swertia japonica (Gentianaceae)⁷ and Spirodela polyrhiza (Lemnaceae).[8,9](#page-3-0) This work used whole plants and did not preclude the possibility of a 2-hydroxyflavanone as the glucosylation substrate. Later work showed incorporation of 2-hydroxyflavanones into C-glucosylflavones by a protein preparation from buckwheat cotyledons, Fagopyrum esculentum (Polygonaceae)^{[10,11](#page-3-0)}

and more recently in wheat, Triticum aestivum L. (Poaceae) and rice, Oryza sativa (Poaceae).^{[12](#page-3-0)}

To study C-glycosylflavonoid biosynthesis in D. uncinatum we completed a total synthesis of 2-hydroxynaringenin (1) using a rearrangement as the key to creating the carbon skeleton ([Scheme](#page-1-0) [1](#page-1-0)). 6-Hydroxy-2,4-dibenzyloxyacetophenone (2) was prepared by benzylation of triacetoxyphloroglucinol¹³ and acylation followed by monodeprotection. Esterification with 4-benzyloxybenzoic acid gave 3, a substrate that yields 4 after the Baker–Venkataraman rearrangement. Deprotection produces 2-hydroxynaringenin (1) which could be further converted into apigenin (5) by dehydration in mild acid and this was hydrogenated to naringenin (6) , with a side product being dihydronaringenin chalcone. NMR analysis of **1** was performed at -60 °C in acetone at which temperature the interconverting ring-closed and open-chain tautomers were dis-tinct on the NMR timescale.^{[14](#page-3-0)} The reaction sequence was repeated using [2,3,5,6⁻²H₄]-4-benzyloxybenzoic acid prepared from $[2,3,4,5,6⁻²H₅]$ -phenol [\(Scheme 2\)](#page-1-0)^{[15](#page-3-0)} to yield the labelled materials $[2',3',5',6'-2H_4]$ -1, $[2',3',5',6'-2H_4]$ -5 and $[2',3',5',6'-2H_4]$ -6.

Incubation of $[2',3',5',6'-^2H_4]$ -1 with UDP-glucose and a protein extract isolated from D. uncinatum leaves was performed.^{16,17} The organic residue was purified by HPLC to afford vitexin (8-C-β-D-glucopyranosylapigenin) and isovitexin (6-C-β-D-glucopyranosylapigenin) which were produced during the assay. Electrospray mass spectrometry showed that during the incubation with leaf protein $[2',3',5',6'-^2H_4]$ -1 was converted into

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Scheme 1. Synthesis of 2-hydroxynaringenin (1). Reagents and conditions: (i) Ac-Q, pyridine; (ii) NaH, BnCl, H-Q, quant. 2 steps; (iii) AcOH, TFAA, 0 °C, 87%; (iv) TiCl₄, 76%; (v) water-soluble DCC, 4-benzyloxybenzoic acid, DMAP; (vi) KOH, pyridine, 75 °C, 67%; (vii) H₂, 10% Pd/C, 65%; (viii) MeOH, few drops 3 N HCl, quant.; (ix) H₂, 10% Pd/C, 25%.

Scheme 2. Synthesis of [2,3,5,6-²H₄]-4-benzyloxybenzoic acid. Reagents and conditions: (i) Br₂, dioxane; (ii) K₂CO₃, BnBr, 63% over 2 steps; (iii) Mg, CO₂, 41%.

both $[2',3',5',6'-^2H_4]$ -vitexin and the regioisomer $[2',3',5',6'-^2H_4]$ isovitexin (Fig. 1). A small quantity of unlabelled vitexin and isovitexin was present in the labelling experiments and protein controls as they are water soluble and co-extracted with the leaf proteins. In further experiments $[2',3',5',6'-^2H_4]$ -5 and $[2',3',5',6'-²H₄]$ -6 were not converted into $[2',3',5',6'-²H₄]$ -vitexin ([Fig. 2](#page-2-0)) and so are not C-glucosylation substrates. There was no conversion of $[2',3',5',6'-^2H_4]$ -1 into vitexin or isovitexin in the controls without the addition of UDP-glucose, nor in experiments containing protein denatured by boiling water, thereby demonstrating glucosylation to be enzyme-mediated.

 $[2',3',5',6'-2H₄]$ -1 exists as a mixture of tautomers and either the open-chain or the ring-closed form may be C-glucosylated. The nucleophilicity of the aromatic ring to facilitate C-glucosylation could be increased by a phenolate ion stabilised by the keto-enol form of the open-chain [[Fig. 3](#page-2-0) (inset)], by analogy to the proposed mechanism of Streptomyces fradiae C-glycosyltransferase, UrdGT2, in which the Asp137 residue removes a phenolic proton, stabilising the substrate and increasing the nucleophilicity.^{[18](#page-3-0)} [2',3',5',6'-²H₄]-1 is a symmetric substrate when in the open-chain form and also after glucosylation, so the production of both C-glucosylflavones, vitexin and isovitexin, may result either from chemical dehydration of the symmetric glucosylated intermediate or by dehydration controlled by the C-glycosyltransferase (CGT) or a CGT-independent dehydratase.

The substrate for C-glucosylation has been demonstrated to be 1 in Poacaea¹² and Polygonaceae,^{[10,11](#page-3-0)} and other work in the Lemnaceae $8,9$ and Gentianaceae^{[7](#page-3-0)} is consistent. We have completed a high yielding total synthesis of 1, which can be adapted to produce substrate analogues, and used stable isotopic labelling of 1 to demonstrate that it is the C-glucosylation substrate for the biosynthesis of C-glucosylflavones vitexin and isovitexin in D. uncinatum (Faba-

Figure 1. ESMS spectra of HPLC-purified vitexin (A) and isovitexin (B) from crude D. uncinatum leaf protein incubated with UDP-glucose and [2',3',5',6'-²H₄]-1.

Figure 2. ESMS spectra of HPLC-purified vitexin from crude D. uncinatum leaf protein incubated with UDP-glucose and (C) [2',3',5',6'-2H₄]-5 or (D) [2',3',5',6'-²H₄]-6.

Figure 3. Biosynthetic pathway for C-glucosylflavonoid biosynthesis in D. uncinatum and (inset) putative mechanism via the keto-enol tautomer of 2-hydroxynaringenin (1).

ceae). This further supports a general biosynthetic pathway for Cglucyosylflavones in plants (Fig. 3).

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- Am. Chem. Soc. 1993, 115, 12296–12304. 16. Materials and methods: Protein isolation: D. uncinatum seeds were obtained
- from Western Seed Company, Kitale, Kenya. Plants were grown in a glasshouse with an 8-h dark/ 16-h light cycle with day and night temperatures of 20 and 18 °C in Rothamsted Standard Compost Mix. Crude protein extract was isolated
according to Brazier and Cole¹⁷ with some modifications. Leaves were harvested, weighed, snap frozen in liquid N_2 and ground to a powder with a pestle and mortar. The powder was extracted on ice with 5 v/w of 0.1 M HEPES–NaOH, pH 7.2, containing 2 mM dithiothreitol (DTT) and 0.025 g/mL of polyvinylpolypyrrolidone (PVP). The mixture was filtered through muslin prior to centrifugation at 8000g for 30 min. Protein was precipitated between 45– 65% (NH4)2SO4 saturation and the pellet recovered by centrifugation at 27000g for 60 min. Protein pellets were frozen in liquid N₂ and stored at –80 °C.

Incubation: Enzyme analysis was performed according to Brazierand Cole 17 with modifications. Protein pellets were desalted in 0.2 M HEPES–NaOH, pH 8, containing 2 mM DTT using Sephadex G-25 gel filtration chromatography (PD-10 columns, GE Healthcare). Protein content was determined using Bradford reagent according to the manufacturer's instructions (Sigma). Protein extract was diluted to 2 mg/mL in buffer and pre-incubated at 30 °C with 66 μ M $[2',3',5',6'-²H₄]$ -1, -5 or -6 for 3 min. The reaction was initiated by the addition of 100 μ M UDP-glucose, stopped after 45 min by the addition of 0.3 M HCl $(250 \,\mu l)$ and extracted with EtOAc, 1.5 mL/mg protein. The organic layer was evaporated under N_2 and re-suspended in MeOH.

HPLC: Samples were isolated on an ACE 5 AQ column (250 mm \times 4.6 mm), using a mobile phase comprising two solvents, solvent A $(H_2O-HCO_2H 95:5)$ and solvent B (MeOH). Samples were eluted with a series of linear gradients set initially at 95:5 A:B (0 min), 85:15 (3 min), 75:25 (13 min), 70:30 (25 min), 45:55 (35 min), 45:55 (45 min), 5:95 (46 min), 5:95 (55 min) and 95:5 (60 min).

ESMS: Purified fractions were collected from the HPLC column, evaporated under N_2 and re-suspended in MeOH prior to analysis by ESMS using an Autospec Ultima Mass Spectrometer [Waters Corp (Micromass UK)] equipped with an electrospray ionisation source (ESI). Positive ion mode ESI-MS was performed at 4 kV and 80 °C, with N₂ gas as the desolvation and nebuliser gas (350 L/h and 15/20 L/h). The solvent used as mobile phase for ESI-MS was $MeOH-H₂O (1:1)$ with 2% AcOH.

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